

Quercetin 3-O- β -D-(6''-caffeoyl)galactoside (1). Yellow needles (200 mg), mp 198–200°, $[\alpha]_D^{20} -19.5^\circ$ (c. 0.12, pyridine), UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (ϵ): 257 (27300), 340 (27900), IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3360, 1650, 1600, 1500, 1353, 1280, 1195, 1160, 1070, FDMS m/z : 627 $[\text{M} + \text{H}]^+$ (100%), 465 (36), ^1H NMR δ (CD_3OD): 5.13 (1H, d, $J = 7.2$ Hz, anomeric), 6.00 (1H, d, $J = 16.2$ Hz, =CH-Ph), 6.13 (1H, d, $J = 1.8$ Hz, H-6), 6.30 (1H, d, $J = 1.8$ Hz, H-8), 6.80 (1H, d, $J = 9$ Hz, H-5'), 7.32 (1H, d, $J = 16.2$ Hz, CO-CH=), 7.55 (1H, dd, $J = 1.8$ and 9 Hz, H-6'), 7.80 (1H, d, $J = 1.8$ Hz, H-2'), 6.7–7.0 (3H, m, aromatic-H of caffeic acid). Found: C, 52.13; H, 4.42. $\text{C}_{30}\text{H}_{26}\text{O}_{15} \cdot 3.5 \text{H}_2\text{O}$ requires C, 52.25; H, 4.82%. 1 (46 mg) was hydrolysed with saturated methanolic Na_2CO_3 (10 ml) at room temp. overnight. The reaction mixture was evaporated *in vacuo* and, after addition of H_2O (5 ml), extracted with EtOAc and then *n*-BuOH. The EtOAc extract gave colourless needles (4.3 mg), mp 220–225°, TLC R_f : 0.82 (TEFF), ^1H NMR δ ($\text{DMSO}-d_6$): 6.21 (1H, d, $J = 16.2$ Hz), 6.74–7.06 (3H, m), 7.46 (1H, d, $J = 16.2$ Hz), identical with caffeic acid. The BuOH extract gave yellow needles (11.4 mg), mp 225–230° of quercetin 3-galactoside [mmp, TLC (EAWA) and NMR].

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KAEMPFEROL 3-SULPHATE IN THE FERN *ADIANTUM CAPILLUS-VENERIS*

FILIPPO IMPERATO

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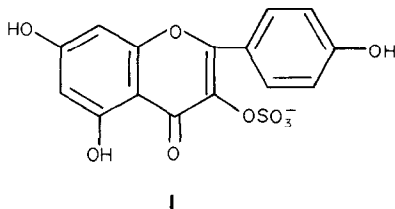
Key Word Index—*Adiantum capillus-veneris*; Adiantaceae; kaempferol 3-sulphate.

Abstract—A new natural product isolated from the fronds of the fern *Adiantum capillus-veneris* has been shown to be kaempferol 3-sulphate by chemical and spectroscopic methods.

Previous work on the phenolic constituents of *Adiantum capillus-veneris* L. has led to the identification of kaempferol 3, 7-diglucoside [1]. Very recently [Imperato, F., unpublished results] four sulphate esters of hydroxycinnamic acid sugar derivatives have been isolated. This paper describes the isolation of kaempferol 3-sulphate from this fern.

This flavonoid (colour reactions: brown to yellow in UV + NH_3) was isolated from an ethanolic extract of fresh fronds of *Adiantum capillus-veneris* by prep.

PC. The UV spectral data: $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 267, 303 (sh), 347; + NaOAc 275, 303, 378; + NaOAc/ H_3BO_3 268, 350; + AlCl_3 278, 304, 349, 396; + AlCl_3 -HCl 276, 303, 343, 395; + NaOMe 275, 322, 392 (increase in intensity); + ZrOCl_2 -citric acid 265, 345 are consistent [2] with those of a 3-substituted flavonol with free hydroxyl groups at positions 5, 7 and 4'. The isolated compound was electrophoretically highly mobile (towards anode), strongly suggestive of a sulphate derivative. Total acid hydrolysis, controlled



acid hydrolysis and alkaline hydrolysis gave kaempferol and sulphate; enzyme hydrolysis with sulphatase (from *Helix pomatia*) gave kaempferol. Kuhn methylation followed by acid hydrolysis gave 5, 7, 4'-tri - O - methylkaempferol. The above results show that the isolated flavonoid must be kaempferol 3-sulphate (1). This compound has been synthesized [3] by Yamaguchi's procedure but is here reported for the first time in plants.

EXPERIMENTAL

Plant material. Fronds of *A. capillus-veneris* L. were collected in Catania.

Isolation of flavonoid. Fresh fronds (100 g) of *A. capillus-veneris* were homogenized and extracted $\times 3$ with 95% EtOH. The combined extracts were filtered, concd. to small vol. *in vacuo* and refiltered. Kaempferol 3-sulphate (15 mg) was isolated by prep. PC on Whatman 3MM paper in BAW; the pale yellow band was cut off, eluted with 70% EtOH, concd. and rechromatographed in 15% HOAc and BEW. R_f data (on Whatman No. 1 paper) are: BAW, 0.66; BEW, 0.28; 15% HOAc, 0.47; PhOH satd. with H_2O , 0.25.

Hydrolysis procedures. Controlled acid hydrolysis was carried out with 10% aq. HOAc (3.5 hr under reflux); total acid hydrolysis was carried out with 2N HCl (2 hr at 100°); alkaline hydrolysis was carried out with 2N NaOH (0.5 hr at 100° under N_2); enzyme hydrolysis with sulphatase was carried out in citrate-Pi buffer, pH 4.5, at 37° for 12 hr. Kaempferol was identified by UV spectroscopy, PC (4 solvents) and polyamide TLC (2 solvents). Sulphate was identified via $BaCl_2$.

Methylation. The flavonoid was methylated with MeI in $HCONMe_2$ in the presence of AgO and the permethylated product was hydrolysed with 0.3 N HCl (4 hr under reflux). 5, 7, 4' - Tri - O - methylkaempferol was identified by UV spectroscopy and PC (2 solvents).

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